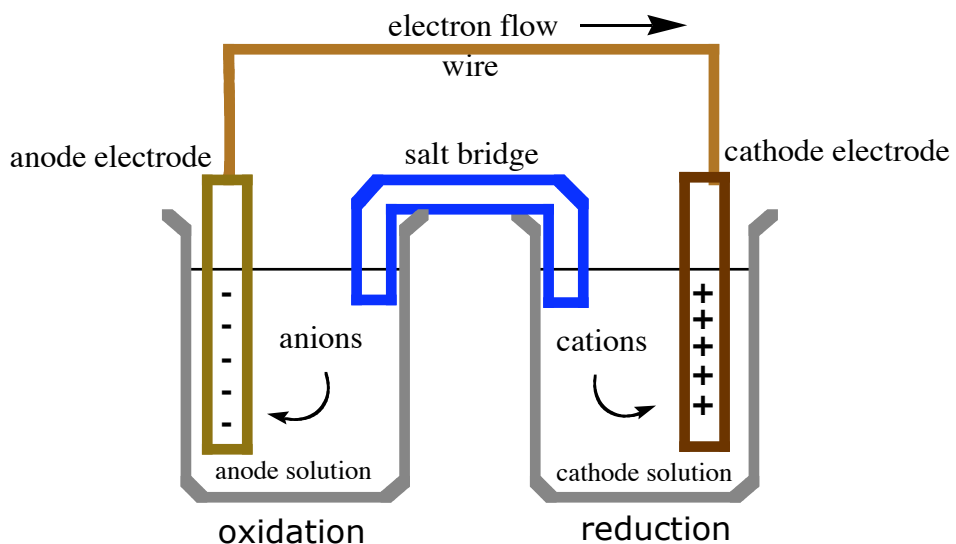


## SIMPLE GALVANIC CELLS



## CELL NOTATION

anode | oxidation reactant | oxidation product || reduction reactant | reduction product | cathode

## STANDARD REDUCTION POTENTIALS

$\text{Cu}^{2+}|\text{Cu}$       +0.337 V

$\text{Sn}^{2+}|\text{Sn}$       -0.14 V

$\text{Pb}^{2+}|\text{Pb}$       -0.126 V

$\text{Fe}^{2+}|\text{Fe}$       -0.44 V

$\text{Zn}^{2+}|\text{Zn}$       -0.763 V